

SAFARI UNIVERSITY, NAIROBI

107/ Kenyatta Ave.

Name of Group: Dr. George Kamukama Department: Electrical Date: 07-05-2014

S/N	Name of member (Full name)	Signature	Time taken (hrs)	Remarks	No. of marks obtained out of 100
1.	Dr. George Kamukama	[Signature]	01:50	Ergonomics Fundamentals	09
2.	[Blank]	[Blank]	02:00	Dynamics Fundamentals	06
3.	[Blank]	[Blank]	01:10	Role of Service Sectors	06
[Blank]	[Blank]	[Blank]	[Blank]	[Blank]	[Blank]
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Small Ergonomics course

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No.	Name of the student	Roll No.	Date	Subject	Marks	Total
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20	...	...	...	...	...	...

Name of the student: ... Roll No.: ... Date: ...  
 Subject: ... Marks: ... Total: ...  
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Name und Matrikelnummer	Matrikelnummer	Vorname	Nachname	Matrikelnummer
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Lab Report

Name of Student: Dr. Prerna Singh Roll No: 19101010101010101010 Date: 10/10/2020

S. No.	Name of the experiment	Theory	Time of time	Steps taken	Precautions
1	Measurement of the length of a rod	Length is the distance between two points.	10 min	1. The rod should be held vertically. 2. The scale should be held horizontally.	1. The eye should be at the level of the scale.
2	Measurement of the area of a rectangle	Area is the space occupied by a closed figure.	10 min	1. The rectangle should be drawn on a white sheet of paper. 2. The length and breadth should be measured.	1. The drawing should be neat and clear.
3	Measurement of the volume of a rectangular block	Volume is the space occupied by a solid.	10 min	1. The block should be held vertically. 2. The length, breadth and height should be measured.	1. The measurements should be taken carefully.
4	Measurement of the mass of a substance	Mass is the quantity of matter contained in a body.	10 min	1. The substance should be weighed in a clean and dry container. 2. The container should be weighed first.	1. The weighing should be done in a draft-free environment.
5	Measurement of the density of a substance	Density is the mass per unit volume of a substance.	10 min	1. The mass and volume of the substance should be measured. 2. The density should be calculated.	1. The measurements should be taken accurately.

Dr. Prerna Singh

Dr. Prerna Singh



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1988-1989

No. of samples collected	Location	Depth	Time	Temperature	Salinity	Density	Other
10	Station 1	0-10m	10:00	15.0	35.0	1.020	Clear
10	Station 2	0-10m	10:15	15.5	35.0	1.020	Clear
10	Station 3	0-10m	10:30	16.0	35.0	1.020	Clear
10	Station 4	0-10m	10:45	16.5	35.0	1.020	Clear
10	Station 5	0-10m	11:00	17.0	35.0	1.020	Clear
10	Station 6	0-10m	11:15	17.5	35.0	1.020	Clear
10	Station 7	0-10m	11:30	18.0	35.0	1.020	Clear
10	Station 8	0-10m	11:45	18.5	35.0	1.020	Clear
10	Station 9	0-10m	12:00	19.0	35.0	1.020	Clear
10	Station 10	0-10m	12:15	19.5	35.0	1.020	Clear

1988-1989



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C.C.	10	15	60	100	150	200	250	300	350	400	450	500	550	600	650	700	750	800	850	900	950	1000
10	15	20	25	30	35	40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120
15	20	25	30	35	40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125
20	25	30	35	40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130
25	30	35	40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135
30	35	40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140
35	40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145
40	45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150
45	50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155
50	55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160
55	60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165
60	65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170
65	70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175
70	75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180
75	80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180	185
80	85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180	185	190
85	90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180	185	190	195
90	95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180	185	190	195	200
95	100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180	185	190	195	200	205
100	105	110	115	120	125	130	135	140	145	150	155	160	165	170	175	180	185	190	195	200	205	210

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Worksheet for Laboratory, 2019-2020

Dr. [Name], [Institution]

Student: [Name] Date: [Date]

No.	Name of the compound	Molecular weight	Boiling point	Solubility	Yield (%)
1	Benzene	78.11	80.1	Insoluble in water, soluble in organic solvents	100
2	Toluene	92.14	110.6	Soluble in water, miscible with organic solvents	100
3	Xylene	106.16	138-140	Soluble in water, miscible with organic solvents	100
4	Ethylbenzene	106.16	106	Soluble in water, miscible with organic solvents	100
5	Styrene	104.12	145	Soluble in water, miscible with organic solvents	100
6	Phenol	94.11	181.7	Soluble in water, miscible with organic solvents	100
7	Aniline	93.09	182	Soluble in water, miscible with organic solvents	100
8	Nitrobenzene	123.11	204	Soluble in water, miscible with organic solvents	100
9	Chlorobenzene	112.56	132	Soluble in water, miscible with organic solvents	100
10	Bromobenzene	157.04	156	Soluble in water, miscible with organic solvents	100

Signature: [Signature]

Date: [Date]

MAHARISHI UNIVERSITY, MUMBAI

10/11/2019

Subject: Physics

Date: 10/11/2019

Sl. No.	Name of the topic / Experiment	Date	Time	Topic / Experiment	No. of marks (out of 10)
1.	Quantum mechanics	10/11/2019	11:00 AM	Quantum mechanics	10
2.	Quantum mechanics	10/11/2019	11:00 AM	Quantum mechanics	10
3.	Quantum mechanics	10/11/2019	11:00 AM	Quantum mechanics	10
4.	Quantum mechanics	10/11/2019	11:00 AM	Quantum mechanics	10
5.	Quantum mechanics	10/11/2019	11:00 AM	Quantum mechanics	10

Dr. T. S. Joshi  
 Head of Department

